

# **Compost or Manure Analysis**

### Sustainable Soil Management with the Mikhail Balance System

DATE ISSUED: 15/09/2025 DATE RECEIVED : 5/09/2025

FILE NO: 2509192740

SAMPLE ID: 50043

LANDTASIA ORGANIC FARMS P/L

PO BOX 116

BUNGENDORE, NSW 2621

REFERENCE PHONE:

CLIENT ID:

REFERENCE:

ANALYSIS REQUIRED :

PHONE:

Total, Available & CEC

LAN055

02 6238 0565

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**Analysis** 

				aryoro	
	ITEM		unit	RESULT	
Basic M	easures:				
	pH (1:5 Water)			8.29	
	pH (1:5 0.01M CaCl <sub>2</sub> )			7.89	
	Electrical Conductivity	EC	μS/cm	1926	
	TOTAL SOLUBLE SALT	TSS	ppm	6356	
Major Nı	utrients:				
					(Major Nutrients in percentages)
	TOTAL NITROGEN	N	kg/t	15.9	1.589 %
	TOTAL PHOSPHORUS	Р	kg/t	2.3	0.2345 %
	TOTAL POTASSIUM	K	kg/t	11.3	1.133 %
	TOTAL SULPHUR	S	kg/t	1.7	0.1663 %
Total Ca	tions:				
	TOTAL CALCIUM	Ca	%	1.58	
	TOTAL MAGNESIUM	Mg	%	0.3587	
	TOTAL SODIUM	Na	%	0.0636	
Trace M	inerals:				
	TOTAL COPPER	Cu	ppm	24.98	
	TOTAL ZINC	Zn	ppm	116	
	TOTAL IRON	Fe	ppm	11621	
	TOTAL MANGANESE	Mn	ppm	478.9	
	TOTAL COBALT	Co	ppm	6.726	
	TOTAL MOLYBDENUM	Мо	ppm	2.016	
	TOTAL BORON	В	ppm	18.08	
Carbon	Content:				
	TOTAL ORGANIC MATTER		%	26	
	TOTAL ORGANIC CARBON		%	13.01	
	CARBON NITROGEN RATIO	C:N		8.2	
	MOISTURE CONTENT	MC	%	57.4	

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# **Plant Available Nutrients**

i lant Available Nutrients					
ITEM		unit	RESULT		
AVAILABLE CALCIUM	Ca	ppm	7402		
AVAILABLE MAGNESIUM	Mg	ppm	1588.8		
AVAILABLE SODIUM	Na	ppm	450.8		
AVAILABLE NITROGEN	N	ppm	73.9		
AVAILABLE PHOSPHORUS	Р	ppm	342.9		
AVAILABLE POTASSIUM	K	ppm	7425.6		
AVAILABLE SULPHUR	S	ppm	81.8		
AVAILABLE COPPER	Cu	ppm	1.52		
AVAILABLE ZINC	Zn	ppm	34.54		
AVAILABLE IRON	Fe	ppm	127.5		
AVAILABLE MANGANESE	Mn	ppm	16.41		
AVAILABLE COBALT	Co	ppm	1.52		
AVAILABLE MOLYBDENUM	Мо	ppm	1.28		
AVAILABLE BORON	В	ppm	4.34		

Notes:

These results represent the proportion of the Total nutrients (page 2) that will be immediately available for plant uptake.

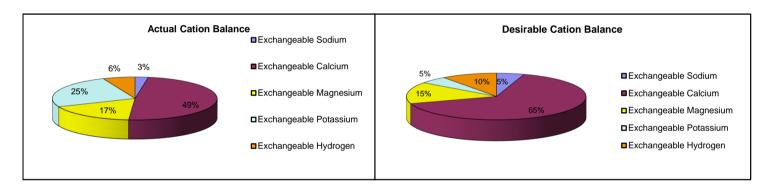
ppm (parts per million) = mg/L (milligram per litre) = mg/kg (milligram per kilogram) 1 % = 10,000 ppm

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# **Exchangeable Cations**

EXCHANGEABLE CATIONS			RESULTS	
CALCIUM	Ca	meq/100g of sample	26.51	
MAGNESIUM	Mg	meq/100g of sample	9.48	
SODIUM	Na	meq/100g of sample	1.40	
POTASSIUM	K	meq/100g of sample	13.64	
HYDROGEN	Н	meq/100g of sample	3.44	
ADJ. EXCH. HYDROGEN	Н	meq/100g of sample	0	
CATION EXCHANGE CAPACITY	CEC	meq/100g of sample	54.47	
ADJUSTED CEC	Adj.CE	EC meq/100g of sample	51.03	
SATURATION BASE PERCENTAGE	BSP		95	
				meq = milliequivalent

<b>EXCHANGEABLE CATION BALANCE</b>		% OF ADJUSTED CEC	DESIRABLE	
CALCIUM PERCENTAGE		51.95	65-70%	
MAGNESIUM PERCENTAGE		18.58	12-15%	
SODIUM PERCENTAGE	ESP	2.74	0.5-5%	
POTASSIUM PERCENTAGE		26.73	3-5%	
ADJ. HYDROGEN PERCENTAGE		0	<20%	
CALCIUM / MAGNESIUM RATIO	Ca/Mg	2.80	2 - 4	



# CATION BALANCE AMENDMENTS (For optimum effectiveness on application)

GYPSUM REQUIREMENT
LIME REQUIREMENT
DOLOMITE REQUIREMENT
MAGNESIUM SULPHATE

2.9 kg/m³

0.0 kg/m³

OR
MAGNESIUM OXIDE
0.0 kg/m³

NB. The effectiveness of the compost may be improved by mixing in the suggested materials (above) prior to application.

### **ANALYTICAL METHODS**

**Methods** 

pH (1:5 Water) 4A1 pH (1:5 CaCl2) 4B1

Electrical conductivity (1:5 Water) 3A1

Total Soluble Salts Calculation from Electrical conductivity

Total Nitrogen

Dumas method, 7A5

Total Calcium, Magnesium, Sodium, Potassium

Acid digestion, ICPAES

Total Phosphorus, Sulphur, Copper, Zinc, Boron

Total Iron, Manganese, Cobalt, Molbydenum

Dumas method, 7A5

Acid digestion, ICPAES

Acid digestion, ICPAES

Exchangeable Calcium, Magnesium, Sodium, Potassium 15D3 or 15A1

Exchangeable Hydrogen Barium Chloride-Triethanolamine method\*

Available Nitrogen Copper-cadmium reductor column at a pH of 8.0

Available Phosphorus Olsen extractable, 9C2a

Available Sulphur KCl 40, 10D1 Available Copper, Zinc, & Cobalt EDTA, 12B1

Available Molybdenum Ammonium Oxalate-Oxalic acid-di-iso propyl ether

Available Iron & Manganese method of E.H. Mikhail (1981)

Available Boron 12C2
Total Organic Carbon Method 6B3
Total Phosphorus, Calcium, Magnesium Acid digestion

NB. For available Iron and Manganese, SWEP uses the method developed by E.H. Mikhail (1980) due to the tendency for the stand ard EDTA method to produce erroneously high results.

For numbered test methods:

Items

Rayment, G.E. & Lyons, D.J. (2011). Soil Chemical Methods - Australasia. CSIRO Publishing, 150 Oxford Street, Collingwood Vic 3066, Australia.

#### AQIS Approved Quarantine Site.

Victorian DPI acccreditation to receive samples from PIZ and PCN infested zones.

**Disclaimer:** All results and/or recommendations in this report are made in good faith and are based on past and ongoing research by SWEP Pty Ltd. However, limitations such as the vagaries of climatic conditions mean that we cannot guarantee production of any crop by the use of this test and associated recommendations, and cannot be held responsible for any results obtained.

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<sup>\*</sup>Peech, M., Cowan, R.L. & Baker, J.H. (1962). Soil Science Society American Procedures, A critical studyof the Barium chloride-Triethanolamine and ammonium acetate methods for determining exchangeable Hydrogen of soils.