

Compost or Manure Analysis

Sustainable Soil Management with the Mikhail Balance System

DATE ISSUED: 15/07/2025 **DATE RECEIVED:** 26/06/2025

CLIENT ID:

REFERENCE:

REFERENCE PHONE:

PHONE:

FILE NO: 2506192445

SAMPLE ID: 50042

LANDTASIA ORGANIC FARMS P/L

PO BOX 116

BUNGENDORE, NSW 2621

BONGENBORE, NOW 202

ANALYSIS REQUIRED: Total, Available

& CEC

LAN055

02 6238 0565

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Analysis

				aryoro	
	ITEM		unit	RESULT	
Basic M	easures:				
	pH (1:5 Water)			8.24	
	pH (1:5 0.01M CaCl ₂)			7.84	
	Electrical Conductivity	EC	μS/cm	1068	
	TOTAL SOLUBLE SALT	TSS	ppm	3524	
Major Nu	utrients:				
					(Major Nutrients in percentages)
	TOTAL NITROGEN	N	kg/t	14	1.4 %
	TOTAL PHOSPHORUS	Р	kg/t	21.4	2.141 %
	TOTAL POTASSIUM	K	kg/t	81.8	8.181 %
	TOTAL SULPHUR	S	kg/t	1.4	0.139 %
Total Ca	tions:				
	TOTAL CALCIUM	Са	%	1.459	
	TOTAL MAGNESIUM	Mg	%	0.303	
	TOTAL SODIUM	Na	%	0.0473	
Trace M	inerals:				
	TOTAL COPPER	Cu	ppm	25.2	
	TOTAL ZINC	Zn	ppm	113	
	TOTAL IRON	Fe	ppm	12290	
	TOTAL MANGANESE	Mn	ppm	410	
	TOTAL COBALT	Co	ppm	5.06	
	TOTAL MOLYBDENUM	Мо	ppm	1.76	
	TOTAL BORON	В	ppm	13.5	
Carbon	Content:				
	TOTAL ORGANIC MATTER		%	49	
	TOTAL ORGANIC CARBON		%	24.5	
	CARBON NITROGEN RATIO	C:N		17.5	
	MOISTURE CONTENT	MC	%	49.4	

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Plant Available Nutrients

i lant Available Nathents					
	unit	RESULT			
Ca	ppm	6126			
Mg	ppm	1200			
Na	ppm	243.8			
N	ppm	58.6			
Р	ppm	262.8			
K	ppm	4305.6			
S	ppm	30.9			
Cu	ppm	1.25			
Zn	ppm	28.93			
Fe	ppm	92.1			
Mn	ppm	9.79			
Co	ppm	1.83			
Мо	ppm	1.13			
В	ppm	3.66			
	Mg Na N P K S Cu Zn Fe Mn Co Mo	Ca ppm Mg ppm Na ppm N ppm P ppm K ppm S ppm Cu ppm Zn ppm Fe ppm Mn ppm Co ppm Mo ppm	unit RESULT Ca ppm 6126 Mg ppm 1200 Na ppm 243.8 N ppm 58.6 P ppm 262.8 K ppm 4305.6 S ppm 30.9 Cu ppm 1.25 Zn ppm 28.93 Fe ppm 92.1 Mn ppm 9.79 Co ppm 1.83 Mo ppm 1.13		

Notes:

These results represent the proportion of the Total nutrients (page 2) that will be immediately available for plant uptake.

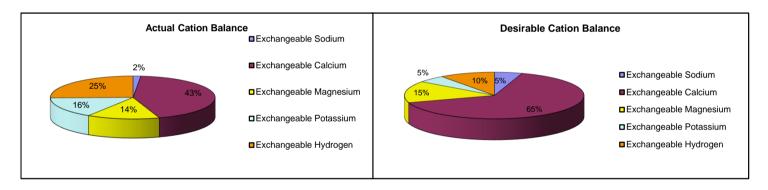
ppm (parts per million) = mg/L (milligram per litre) = mg/kg (milligram per kilogram) 1 % = 10,000 ppm

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Exchangeable Cations

EXCHANGEABLE CATIONS			RESULTS	
CALCIUM	Ca	meq/100g of sample	25.30	
MAGNESIUM	Mg	meq/100g of sample	8.26	
SODIUM	Na	meq/100g of sample	0.88	
POTASSIUM	K	meq/100g of sample	9.12	
HYDROGEN	Н	meq/100g of sample	14.80	
ADJ. EXCH. HYDROGEN	Н	meq/100g of sample	0	
CATION EXCHANGE CAPACITY	CEC	meq/100g of sample	58.36	
ADJUSTED CEC	Adj.CE	C meq/100g of sample	43.56	
SATURATION BASE PERCENTAGE	BSP		78	
				meq = milliequivalent

EXCHANGEABLE CATION BALANCE		% OF ADJUSTED CEC	DESIRABLE	
CALCIUM PERCENTAGE		58.08	65-70%	
MAGNESIUM PERCENTAGE		18.96	12-15%	
SODIUM PERCENTAGE	ESP	2.02	0.5-5%	
POTASSIUM PERCENTAGE		20.94	3-5%	
ADJ. HYDROGEN PERCENTAGE		0	<20%	
CALCIUM / MAGNESIUM RATIO	Ca/Mg	3.06	2 - 4	



CATION BALANCE AMENDMENTS (For optimum effectiveness on application)

GYPSUM REQUIREMENT	2.8 kg/m ³			
LIME REQUIREMENT	0.0 kg/m ³			
DOLOMITE REQUIREMENT	0.0 kg/m ³			
MAGNESIUM SULPHATE	0.0 kg/m ³	OR	MAGNESIUM OXIDE	0.0 kg/m ³

NB. The effectiveness of the compost may be improved by mixing in the suggested materials (above) prior to application.

ANALYTICAL METHODS

Methods

pH (1:5 Water) 4A1 pH (1:5 CaCl2) 4B1

Electrical conductivity (1:5 Water) 3A1

Total Soluble Salts Calculation from Electrical conductivity

Total Nitrogen

Dumas method, 7A5

Total Calcium, Magnesium, Sodium, Potassium

Acid digestion, ICPAES

Total Phosphorus, Sulphur, Copper, Zinc, Boron

Total Iron, Manganese, Cobalt, Molbydenum

Dumas method, 7A5

Acid digestion, ICPAES

Acid digestion, ICPAES

Exchangeable Calcium, Magnesium, Sodium, Potassium 15D3 or 15A1

Exchangeable Hydrogen Barium Chloride-Triethanolamine method*

Available Nitrogen Copper-cadmium reductor column at a pH of 8.0

Available Phosphorus Olsen extractable, 9C2a

Available Sulphur KCl 40, 10D1 Available Copper, Zinc, & Cobalt EDTA, 12B1

Available Molybdenum Ammonium Oxalate-Oxalic acid-di-iso propyl ether

Available Iron & Manganese method of E.H. Mikhail (1981)

Available Boron 12C2
Total Organic Carbon Method 6B3
Total Phosphorus, Calcium, Magnesium Acid digestion

NB. For available Iron and Manganese, SWEP uses the method developed by E.H. Mikhail (1980) due to the tendency for the stand ard EDTA method to produce erroneously high results.

For numbered test methods:

Items

Rayment, G.E. & Lyons, D.J. (2011). Soil Chemical Methods - Australasia. CSIRO Publishing, 150 Oxford Street, Collingwood Vic 3066, Australia.

AQIS Approved Quarantine Site.

Victorian DPI acccreditation to receive samples from PIZ and PCN infested zones.

Disclaimer: All results and/or recommendations in this report are made in good faith and are based on past and ongoing research by SWEP Pty Ltd. However, limitations such as the vagaries of climatic conditions mean that we cannot guarantee production of any crop by the use of this test and associated recommendations, and cannot be held responsible for any results obtained.

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^{*}Peech, M., Cowan, R.L. & Baker, J.H. (1962). Soil Science Society American Procedures, A critical studyof the Barium chloride-Triethanolamine and ammonium acetate methods for determining exchangeable Hydrogen of soils.